(19) World Intellectual Property Organization International Bureau



(43) International Publication Date 9 January 2003 (09.01.2003)

PCT

(10) International Publication Number WO 03/002674 A1

- (51) International Patent Classification⁷: C09D 5/36, 11/00, C08J 7/04, B05D 7/00, B44C 1/17, A61K 7/42, C03C 17/32, C09D 131/04
- (21) International Application Number: PCT/US02/20870
- (22) International Filing Date: 27 June 2002 (27.06.2002)
- (25) Filing Language:

English

(26) Publication Language:

English

(30) Priority Data: 60/301,276 2

27 June 2001 (27.06.2001) US

-) US
- (71) Applicant: PENNZOIL-QUAKER STATE COMPANY [US/US]; 700 Milam, P.O. Box 2967, Houston, TX 77002 (US).
- (72) Inventors: SITABKHAN, Arif, M.; 3718 Cinnamon Way, Westlake, OH 44145 (US). HOWE, Michael, W.; 111 N. Taylor Point Dr., The Woodlands, TX 77382 (US). MATHEWS, Roderic; 64 Hickory Oak Dr., The Woodlands, TX 77381 (US). FANG, Jiafu; 2707 Honeysuckle Walk, Spring, TX 77388 (US). BRUN, Anne, M.; 58 Lightwood Trace, The Woodlands, TX 77382 (US).
- (74) Agent: BAI, J., Benjamin; Jenkens & Gilchrist, 1100 Louisiana, Ste. 1800, Houston, TX 77002 (US).

- (81) Designated States (national): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZM, ZW.
- (84) Designated States (regional): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

Published:

- with international search report
- before the expiration of the time limit for amending the claims and to be republished in the event of receipt of amendments

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

(54) Title: COATING COMPOSITION

(57) Abstract: A tinting composition for tinting a surface which includes a solvent, a film former, and a pigment, such as iron oxide, titanium dioxide, mica, calcium sodium borosilicate, organic constituent containing pigments, inorganic constituent containing pigments, fluorescent constituent containing pigments, phosphorescent constituent containing pigments, luminescent constituent containing pigments, ultraviolet absorbing pigments, mineral constituent containing pigments, amorphous glass oxides, polyacrylates, metal flakes, reflective pigments, and mixtures thereof. A method of tinting a surface which includes selecting a surface for tinting; mixing a solvent, a film former and a pigment to form a tinting composition; and applying the tinting composition to the surface.



COATING COMPOSITION

5

10

15

20

25

FIELD OF THE INVENTION

This invention relates to a composition for tinting surfaces and methods of using the composition. More particularly, the invention relates to a composition for tinting polymeric substrates such as rubber, engineered plastics, and vinyl, metals, glass, fiberglass, paint, and plastics and methods of using the composition.

BACKGROUND OF THE INVENTION

Increasingly, consumers are expending money on accessories for their vehicles. Consumers often wish to modify the surface color of accessories used on or inside their vehicle to either coordinate or contrast that accessory with other parts of their vehicle or to accentuate that color. Examples of accessories that consumers may wish to modify in color to coordinate or contrast with other parts of their vehicle or to accentuate include, but are not limited to, polymeric substrates such as rubber, engineered plastic, and vinyl accessories, metal accessories, glass accessories, fiberglass accessories, painted accessories, and plastic accessories.

Examples of polymeric substrates such as rubber and vinyl accessories on vehicles that consumers may wish to modify in color include, but are not limited to, tires, dashboards, bumpers, seats, steering wheels, and mudflaps. Engineered plastic accessories on vehicles that consumers may wish to modify in color include, but are not limited to, bumpers, body panels, trim, and consoles. Metal accessories on vehicles that consumers may wish to modify in color include, but are not limited to, bumpers, rims, grills and hitches. Also, glass accessories on vehicles that consumers may wish to modify in color include, but are not limited to, windshields and mirrors. Certain fiberglass accessories on vehicles that consumers may wish to modify in color include, but are not limited to, body panels. Painted accessories on vehicles that consumers may wish to modify in color include, but are not limited to, exterior vehicle paint. Certain

2

plastic accessories on vehicles that consumers may wish to modify in color include, but are not limited to, bumpers, headlights, and mirror casings.

Although consumers may wish to alter the surface color of the accessories used on or inside their vehicles to coordinate or contrast that color with other parts of their vehicle or to accentuate that color, often times consumers only desire to make the color change on a temporary, nonpermanent basis. Accordingly, a need exists for a tinting composition which consumers can safely handle and apply to accessories on or inside their vehicles or other surfaces not necessarily associated with vehicles to temporarily impart a color of their choice to the surface of the accessory.

10

15

5

SUMMARY OF THE INVENTION

The embodiments of the invention meet the above need by providing a tinting composition which includes a solvent, a film former, and a pigment. In some embodiments, the composition does not include (i.e., substantially free of) a pigment and thus is clear. It can form a protective coating on the surface. The composition may be applied to a variety of surfaces including tires, glass, and plastic. The composition may be applied to the surface by sponging, dabbing, painting and wiping the tinting composition onto the surface.

BRIEF DESCRIPTION OF THE DRAWINGS

Not applicable.

20

25

DESCRIPTION OF EMBODIMENTS OF THE INVENTION

Embodiments of the invention provide a new tinting composition that is capable of temporarily tinting a surface, for example, the exterior of an automobile. The tinting composition comprises a solvent, a film former, and a pigment. Other optional additives may also be used. In some embodiments, the composition includes water; in other embodiments, the composition is substantially free of water. Although preferred uses of the tinting composition are directed to vehicles and vehicle accessories, other surfaces can be tinted. Examples of suitable surfaces which can be tinted by the tinting

3

composition include, but are not limited to, polymeric substrates such as rubber, engineered plastics, and vinyl, metal, glass, fiberglass, painted surfaces, and plastic surfaces.

5

10

15

20

25

In some embodiments, the solvent is selected from the group consisting of branched or straight chained aliphatic hydrocarbons, synthetic hydrocarbons, alcohols, water, organic solvents, silicone fluids, paraffinic solvents, volatile solvents, and mixtures thereof. The pigment is selected from the group consisting of iron oxide, titanium dioxide, mica, calcium sodium borosilicate, organic constituent containing pigments, inorganic constituent containing pigments, fluorescent constituent containing pigments, phosphorescent constituent containing pigments, luminescent constituent containing pigments, ultraviolet absorbing pigments, mineral constituent containing pigments, amorphous glass oxides, polyacrylates, metal flakes, reflective pigments, or mixtures thereof.

Embodiments of the invention also provide a new method of tinting a surface. The method includes selecting a surface for tinting, mixing a solvent, a film former and a pigment to form a tinting composition, and applying the tinting composition to the surface and optionally includes the step of drying the tinting composition. The solvent is selected from the group consisting of branched or straight chained aliphatic hydrocarbons, synthetic hydrocarbons, alcohols, water, organic solvents, silicone fluids, paraffinic solvents, volatile solvents, and mixtures thereof. The pigment is selected from the group consisting of iron oxide, titanium dioxide, mica, calcium sodium borosilicate, organic constituent containing pigments, inorganic constituent containing pigments, fluorescent constituent containing pigments, phosphorescent constituent containing pigments, ultraviolet absorbing pigments, mineral constituent containing pigments, amorphous glass oxides, polyacrylates, metal flakes, reflective pigments, or mixtures thereof.

When applied to the surface, the tinting composition is liquid in form. The tinting composition can be applied in various manners to the surface. It is contemplated that the tinting composition may be applied to the surface by spraying the tinting composition

4

onto the surface with a spray trigger or an aerosol can; by sponging, dabbing, painting, or wiping the tinting composition onto the surface with a cloth, brush, sponge, pen, marker, or similar application means; by applying an adhesive which contains the tinting composition to the surface and then removing the adhesive; or by any other suitable means for applying the tinting composition to the surface. Once the tinting composition has been applied to the surface, the tinting composition dries and imparts a color and sheen to the surface. The drying time depends upon the length of time required for the solvent in the tinting composition to evaporate and is affected by the ventilation and the ambient conditions such as the humidity level and the temperature. The drying time can vary from about 10 minutes to about an hour depending on the ambient conditions and the solvent selected. The drying can be accomplished by allowing the tinting composition to be exposed to the ambient conditions or by applying heat to the tinting composition.

Once the tinting composition has been applied to the surface, the composition will remain on the surface for days to months. The length of time that the tinting composition remains on the surface depends upon a variety of conditions such as the ambient and environmental conditions, the amount of tinting composition which has been applied to the surface, the driving conditions to which the tinting composition is subjected, and the amount of abrasion to which the tinting composition is subjected. The tinting composition will eventually fade and/or peel and will need to be reapplied to the surface should the consumer wish to again impart a tint to the surface. Should the consumer desire to remove the tinting composition from the surface before it fades and/or peels, one can use a cleaning solution and optionally a brush to remove the tinting composition. In some embodiments, the tinting composition, after drying on a substrate, can be easily peeled by hand. In other words, the tinting composition forms a peelable coating after being applied to a substrate. In some cases, one or more peeling aids are used to improve the peelability of the coating.

The selection of the solvent for use in the tinting composition depends on a variety of factors such as the desired drying time of the tinting composition, the

5

0.

5

0.

!5

5

flowability of the tinting composition, the desired degree of dispersing of the pigment in the tinting composition, and environmental considerations. Generally, the solvent is present in the tinting composition in an amount from about 0.1 to about 99.9 wt.% of the tinting composition. In certain embodiments, the solvent is present in an amount from about 50 to about 99.9 wt.% of the tinting composition. In certain other embodiments, the solvent is present in an amount from about 75 to about 99.9 wt.% of the tinting composition.

5

10

15

20

25

Examples of suitable solvents for use in the tinting composition include, but are not limited to, branched or straight chained aliphatic hydrocarbons, synthetic hydrocarbons, alcohols, water, organic solvents, silicone fluids, paraffinic solvents, volatile solvents, and mixtures thereof. Examples of suitable branched or straight chained aliphatic hydrocarbons include, but are not limited to, Shell Sol 71 and Shell Sol 142HT made by Shell Chemical Company, 2251 Oil made by Penreco, and Conosol® C145 made by Penreco. Examples of suitable synthetic hydrocarbons include, but are not limited to, synthetic isoparaffinic hydrocarbons such as Isopar® M made by ExxonMobil Chemical Company. Examples of suitable alcohols include, but are not limited to, methanols, ethanols, propanols, and tert-butyl alcohols. Examples of suitable organic solvents include acetone, hexane, toluene, esters, volatile aldehydes, and volatile ketones. Examples of suitable silicone fluids include, but are not limited to, polydimethylsiloxane, functional silicone polymers, silicone resins, volatile silicones, and silanes. Examples of suitable functional silicone polymers include, but are not limited to, silicones containing amine functional groups, fluorinated functional groups, hydroxy terminated functional groups, cyclic functional groups, vinyl functional groups, and ethoxy functional groups. Where silicone fluids are used as the solvent in the tinting composition, the preferred viscosity range is from about 0.65 to about 12,000,000 cst. Examples of suitable paraffinic solvents include, but are not limited to, Isopar® M made by ExxonMobil Chemical Company. Examples of suitable volatile solvents include, but are not limited to, cyclic silicones, acetone, and mineral spirits.

6

Suitable solvents include, but are not limited to the following compounds identified by their INCI names: 1,2,6-Hexanetriol; 3-Hexenol; Acetone; Alcohol; Alcohol Denat.; Amyl Acetate; Aqua; Arachis Hypogaea; Benzaldehyde; Benzyl Alcohol; Benzyl Benzoate; Bertholletia Excelsa; Brevoortia; Bubulum; Butoxydiglycol; Butoxyethanol; Butyl Acetate; Butylene Glycol; Butyrolactone; C₁₀₋₁₁ Isoparaffin; C₁₀₋₁₃ Isoparaffin; C₁₀₋₁₈ Triglycerides; C₁₁₋₁₂ Isoparaffin; C₁₁₋₁₃ Isoparaffin; C₁₂₋₁₄ Isoparaffin; C₁₃₋₁₄ Isoparaffin; C₁₃₋₁₆ Isoparaffin; C₂₀₋₄₀ Isoparaffin; C₇₋₈ Isoparaffin; C₈₋₉ Isoparaffin; C₉₋₁₁ Isoparaffin; C₉₋₁₃ Isoparaffin; C₉₋₁₄ Isoparaffin; Caprylic / Capric / Lauric Triglyceride; Caprylic / Capric / Stearic Triglyceride; Caprylic / Capric Triglyceride; Cocos Nucifera; Coenzyme A; Cyclohexane; Cyclomethicone; Deodorized Kerosene; Diacetin; Diacetone Alcohol; Dibutyl Phthalate; Dichloromethane; Diethoxyethyl Succinate; Diethyl Phthalate; Diethylene Glycol; Dimethoxydiglycol; Dimethyl Ether; Dimethyl Isosorbide; Dimethyl Phthalate; Dimethyl Sulfone; Dipentene; Diphenyl Methane; Dipropylene Glycol; Ethoxydiglycol; Ethoxydiglycol Acetate; Ethoxyethanol; Ethoxyethanol Acetate; Ethyl Acetate; Ethyl Hexanediol; Ethylene Carbonate; Furfural; Glycereth-12; Glycereth-26; Glycereth-7; Glycerin; Glyceryl Triacetyl Hydroxystearate; Glyceryl Triacetyl Ricinoleate; Glycofurol; Glycol; Heptane; Hexane; Hexyl Alcohol; Hexyldecanol; Hexylene Glycol; Hydrogenated Menhaden Oil; Hydroxyethyl Pei-1000; Hydroxyethyl Pei-1500; Isoamyl Acetate; Isobutoxypropanol; Isobutyl Acetate; Isocetyl Salicylate; Isohexadecane; Isopentane; Isopropyl Acetate; Isopropyl Alcohol; Isopropyl Myristate; Isopropyl Palmitate; Isopropyl Ppg-2-Isodeceth-7 Carboxylate; Lanolin; Mek; Methoxydiglycol; Methoxyethanol; Methoxyisopropanol; Methoxymethylbutanol; Methyl Acetate; Methyl Alcohol; Methyl Hexyl Ether; Methylal; Methylpropanediol; Mibk; Mineral Spirits; Mixed Terpenes; N-Butyl Alcohol; Octyldodecanol; Olea Europaea; Papaver Orientale; Paraffinum Liquidum; PEG / PPG -17 / 6 Copolymer; PEG / PPG-18 / 4 Copolymer; PEG / PPG -23 / 50 Copolymer; PEG / PPG -296 / 57 Copolymer; PEG-10; PEG -12; PEG -135; PEG -14; PEG -150; PEG -16; PEG -18; PEG -20; PEG -200; PEG -32; PEG -350; PEG -4; PEG -40; PEG -6; PEG -6 Methyl Ether; Peg-75; PEG -8; PEG -9; PEG -90; Pelargonic Acid; Pentane; Petroleum Distillates;

5

10

15

20

7

Phenoxyisopropanol; Phenylpropanol; PPG-2 Methyl Ether; PPG-3 Methyl Ether; Propyl Acetate; Propyl Alcohol; Propylene Carbonate; Propylene Glycol; Sorbeth-20; T-Butyl Alcohol; Tallol; Taraktogenos Kurzii; Terpineol; Tetrahydrofurfuryl Tetrahydrofurfuryl Alcohol; Thiolanediol; Toluene; Triacetin; Tributyl Citrate; Tributylcresylbutane; Tricaprin; Trichloroethane; Triethyl Citrate; Triethylene Glycol; Trilinolein; Trilaurin; Triisostearin; Triisononanoin; Trihydroxystearin; Tripalmitin; Tristearin; Triolein; Trioctanoin; Trimyristin; Trimethylhexanol; Triundecanoin; Turpentine; Xylene; and Zea Mays.

5

10

15

20

25

As mentioned above, the tinting composition comprises one or more film formers. Any film formers may be used in embodiments of the invention. Suitable film formers include, but are not limited to the following compounds identified by their respective Acrylamides/ copolymer; acrylate Acrylamide/sodium INCI names: acrylates/DMAPA/methoxy PEG methacrylate copolymer; Acrylamides copolymer; Acrylamidopropyltrimonium chloride/acrylates copolymer; Acrylates/ acetoacetoxyethyl methacrylate copolymer; Acrylates/acrylamide copolymer; Acrylates/ ammonium methacrylate copolymer; Acrylates/C₁₀₋₃₀ alkyl acrylate crosspolymer; Acrylates/ diacetoneacrylamide copolymer; Acrylates/octylacrylamide copolymer; Acrylates/PVP copolymer; Acrylates/ steareth-50 acrylate copolymer; Acrylates/VA copolymer; copolymer; Acrylic acid/acrylonitrogens Acrylates/VA crosspolymer; Acrylates glycol/glycerin crosspolymer; Adipic diethylene Adipic acid/ copolymer; acid/dimethylaminohydroxypropyl Adipic acid/diethylenetriamine copolymer; diethylenetriamine copolymer; Adipic acid/epoxypropyl diethylenetriamine copolymer; Adipic acid/isophthalic acid/neopentyl glycol/trimethylolpropane copolymer; Albumen; Allyl stearate/VA copolymer; Aminoethylacrylate phosphate/acrylates copolymer; Ammonium acrylates/acrylonitrogens copolymer; Ammonium acrylates copolymer; copolymer; VA/acrylates Ammonium alginate; Ammonium copolymer; AMP-acrylates AMP-acrylates/diacetoneacrylamide copolymer; AMPD-acrylates/diacetoneacrylamide copolymer; Bayberry wax; Behenyl/isostearyl beeswax; Benzoic acid/phthalic anhydride/pentaerythritol/ neopentyl glycol/palmitic acid

8

copolymer; Butadiene/acrylonitrile copolymer; Butoxy chitosan; Butyl benzoic acid/phthalic anhydride/trimethylolethane copolymer; Butyl benzyl phthalate; Butyl ester of ethylene/MA copolymer; Butyl ester of PVM/MA copolymer; Butyl phthalyl butyl glycolate; Butylated polyoxymethylene urea; Butylated PVP; Calcium/sodium PVM/MA copolymer; Calcium carrageenan; Candelilla cera; Carboxymethyl chitosan succinamide; Carboxymethyl hydroxyethylcellulose; Carnauba; Cellulose acetate; Cellulose acetate butyrate; Cellulose acetate propionate; Cellulose gum; Cetyl hydroxyethylcellulose; Chitosan succinamide; Collodion; Corn starch/acrylamide/sodium acrylate copolymer; Dilinoleic acid/ ethylenediamine copolymer; Dimethicone/mercaptopropyl methicone copolymer; Dimethicone/sodium PG-propyldimethicone thiosulfate copolymer; Dimethyl DMAPA acrylates/acrylic acid/acrylonitrogens copolymer; DMHF; phthalate; Dodecanedioic acid/cetearyl alcohol/glycol copolymer; Ethyl ester of PVM/MA copolymer; Ethylcellulose; Ethylene/acrylic acid/VA copolymer; Ethylene/acrylic acid copolymer; Ethylene/MA copolymer; acrylate Ethylene/calcium copolymer; Ethylene/VA copolymer; Glyceryl alginate; Glyceryl hydrogenated rosinate; Glyceryl polyacrylate; Glyceryl rosinate; Glycosaminoglycans; Guar hydroxypropyltrimonium chloride; Gutta percha; Hydrogenated styrene/methyl styrene/indene copolymer; collagen; Hydrolyzed elastin; Hydrolyzed keratin; Hydroxybutyl Hydrolyzed methylcellulose; Hydroxyethyl ethylcellulose; Hydroxyethylcellulose; Hydroxylated lanolin; Hydroxypropyl guar; Hydroxypropyl methylcellulose; Hydroxypropylcellulose; Isobutylene/ sodium maleate copolymer; Isopropyl ester of PVM/MA copolymer; Lanolin cera; Lauryl acrylate/VA copolymer; Maltodextrin; Methacryloyl ethyl betaine/acrylates copolymer; Methyl hydrogenated rosinate; Methyl methacrylate crosspolymer; Methyl rosinate; Nonoxynyl hydroxyethylcellulose; Oat beta glucan; Octylacrylamide/acrylates/ butylaminoethyl methacrylate copolymer; Oleoyl hydrolyzed collagen; PEG-8/SMDI copolymer; PEG-crosspolymer; Pentaerythrityl hydrogenated rosinate; Pentaerythrityl rosinate; Phthalic anhydride/adipic acid/castor oil/neopentyl Phthalic copolymer; anhydride/benzoic glycol/PEG-3/trimethylolpropane acid/trimethylolpropane copolymer; Phthalic anhydride/butyl benzoic acid/propylene

5

10

15

20

9

5

10

15

20

25

glycol copolymer; Phthalic anhydride/glycerin/glycidyl decanoate copolymer; Phthalic Polyacrylamide; anhydride/glycols copolymer; anhydride/trimellitic Polybutylene acid; sulfonic acid; Polyacrylic Polyacrylamidomethylpropane methacrylate; Polyglucuronic acid; Polydimethylaminoethyl terephthalate; Polyglycerylmethacrylate; Polymethacrylamidopropyltrimonium chloride; Polymethyl acrylate; Polymethyl methacrylate; Polyoxyisobutylene/methylene urea copolymer; Polyquaternium-12; Polyquaternium-11; Polyquaternium-1; Polyquaternium-10; Polyquaternium-16; Polyquaternium-15; Polyquaternium-14; Polyquaternium-13; Polyquaternium-2; Polyquaternium-19; Polyquaternium-17; Polyquaternium-18; Polyquaternium-27; Polyquaternium-24; Polyquaternium-22; Polyquaternium-20; Polyquaternium-31; Polyquaternium-30; Polyquaternium-29; Polyquaternium-28; Polyquaternium-35; Polyquaternium-34; Polyquaternium-33; Polyquaternium-32; Polyquaternium-4; Polyquaternium-39; Polyquaternium-37; Polyquaternium-36; Polyquaternium-7; Polyquaternium-5; Polyquaternium-6; Polyquaternium-42; Polyquaternium-9; Polysilicone-6; Polystyrene; Polyurethane; Polyquaternium-8; Polyvinyl acetate; Polyvinyl alcohol; Polyvinyl butyral; Polyvinyl imidazolinium acetate; Polyvinyl laurate; Polyvinyl methyl ether; Potassium carrageenan; Potassium hyaluronate; PPG-26/TDI copolymer; PPG-51/SMDI copolymer; Procollagen; PVM/MA copolymer; PVP; PVP/decene copolymer; PVP/dimethylaminoethylmethacrylate copolymer; PVP/eicosene copolymer; PVP/hexadecene copolymer; PVP/VA/itaconic acid copolymer; PVP/VA/vinyl propionate copolymer; PVP/VA copolymer; Rosin acrylate; Rosin hydrolyzed collagen; Shellac; Shellac cera; Sodium acrylate/vinyl alcohol copolymer; Sodium carrageenan; Sodium DVB/acrylates copolymer; Sodium polyacrylate starch; Sodium polymethacrylate; Sodium polystyrene sulfonate; Sodium PVM/MA/decadiene crosspolymer; Sodium styrene/acrylamide copolymer; Sodium styrene/acrylates copolymer; Sodium tauride acrylates/acrylic acid/acrylonitrogens Starch/acrylates/acrylamide copolymer; Starch collagen; Soluble copolymer; diethylaminoethyl ether; Steareth-10 allyl ether/acrylates copolymer; Stearylvinyl ether/MA copolymer; Styrene/acrylates/acrylonitrile copolymer; Styrene/acrylates/ 5

10

15

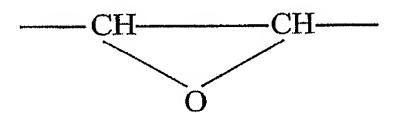
20

25

ammonium methacrylate copolymer; Styrene/allyl benzoate copolymer; Styrene/MA copolymer; Styrene/PVP copolymer; Sucrose acetate isobutyrate; Sucrose benzoate; Sucrose benzoate/ sucrose acetate isobutyrate/butyl benzyl phthalate/methyl methacrylate copolymer; Sucrose benzoate/sucrose acetate isobutyrate/butyl benzyl phthalate copolymer; isobutyrate sucrose acetate benzoate/ Sucrose copolymer; PVP; Tricontanyl Tea-acrylates/acrylonitrogens copolymer; Trimethylpentanediol/isophthalic acid/trimellitic anhydride copolymer; Tromethamine acrylates/acrylonitrogens copolymer; VA/butyl maleate/isobornyl acrylate copolymer; VA/crotonates/methacryloxybenzophenone-1 VA/crotonates/vinyl copolymer; neodecanoate copolymer; VA/crotonates/vinyl propionate copolymer; VA/crotonates copolymer; VA/DBM copolymer; VA/isobutyl maleate/vinyl neodecanoate copolymer; VA/vinyl butyl benzoate/ crotonates copolymer; Vinyl acetate; and Vinyl caprolactam/PVP/dimethylaminoethyl methacrylate copolymer.

In some embodiments, the film-forming resin can be an epoxy, vinyl, alkyd, polyester, acrylic, aminoplast, phenolplast, cellulose derivative, amide or urethane resin or mixtures thereof. Copolymers derived from such resins are also useful herein.

The epoxide resins used as a film-forming resin in the tinting compositions are those compounds having a 1,2-epoxy group, i.e.,



present in the molecule. Polyepoxides contain more than one 1,2-epoxy group per molecule. In general, the epoxide equivalent weight will range from about 140 to about 4,000. These polyepoxides are saturated or unsaturated, cyclic or acyclic, aliphatic, alicyclic, aromatic or heterocyclic. They can contain substituents such as halogen, hydroxyl and ether groups.

One useful class of polyepoxides comprises the epoxy polyethers obtained by reacting an epihalohydrin (such as epichlorohydrin or epibromohydrin) with a polyphenol

11

in the presence of an alkali. Suitable polyphenols include resorcinol, catechol, hydroquinone, bis(4-hydroxyphenyl)-2,2-propane, i.e., bisphenol A; bis(4-hydroxyphenyl)-1,1-isobutane;4,4-dihydroxybenzophenone; bis(4-hydroxyphenyl)-1,1-ethane; bis(2-hydroxynaphenyl)-methane; and 1,5-hydroxynaphthalene. One very common polyepoxide is a polyglycidyl ether of a polyphenol, such as bisphenol A.

5

10

15

20

Another class of epoxy resins are the polyglycidyl ethers of polyhydrix alcohols. These compounds may be derived from such polyhydric alcohols as ethylene glycol, diethylene glycol, triethylene glycol, 1,2-propylene glycol, 1,4-butylene glycol, 1,5-pentanediol, 1,2,6-hexanetriol, glycerol, trimethylolpropane, and bis(4-hydroxycyclohexyl)-2,2-propane.

Another class of epoxide resins are the polyglycidyl esters of polycarboxylic acids. These compounds are produced by the reaction of epichlorohydrin or a similar epoxy compound with an aliphatic or aromatic polycarboxylic acid, such as oxalic acid, succinic acid, glutaric acid, terephthalic acid, 2,6-naphthalene dicarboxylic acid and dimerized linoleic acid.

Still another class of polyepoxides are derived from the epoxidation of an olefinicially unsaturated alicyclic compound. These polyepoxides are non-phenolic and are obtained by epoxidation of alicyclic olefins, for example, by oxygen and selected metal catalysts, by perbenzoic acid, by acid-aldehyde monoperacetate or by peracetic acid. Among such polyepoxides are the epoxy alicyclic ethers and esters well known in the art.

Useful polyepoxides also include those containing oxyalkylene groups in the epoxy molecule. Such oxyalkylene groups have the general formula:

where R is hydrogen or alkyl, preferably a lower alkyl having from 1 to 6 carbon atoms, m is 1 to 4 and n is 2 to 50. Such groups are pendant to the main molecular chain of the polyepoxide or are part of the main chain itself. The proportion of oxyalkylene groups in the polyepoxide depends upon many factors, including the chain length of the oxyalkylene group, the nature of the epoxy and the degree of water solubility desired.

Another class of polyepoxides consists of the epoxy novolac resins. These resins are obtained by reacting an epihalohydrin with the condensation product of aldehyde and monohydric or polyhydric phenols. A typical example is the reaction product of epichlorohydrin with a phenolformaldehyde condensate.

Any well-known curing reactant for the above-described epoxy resin is normally included in the tinting composition. It is well known, chemicals and resins containing functional groups with active hydrogen groups are useful as curing agents for the epoxy resins. Generally, the curing agents cause polymerization by cross-linking of the epoxy molecules. Amine and polyamide catalysts are especially preferred curing agents.

Vinyl resins used in the tinting compositions are derived from monomers containing a carbon to carbon double bond. These monomers polymerize by linear addition to form long chain molecules. Generally, the polymeric resins have the structure:

where R and R₁ represent various pendant groups such as hydrogen, chlorine, acetate, benzene and toluene. The vinyl resins are commonly derived from the monomers vinyl chloride, vinylidene chloride, vinyl acetate, the vinyl acetals, styrene, acrylonitrile and mixtures thereof. The vinyl polymers and copolymers range from about 100 to 10,000

10

5

15

13

carbon atoms in chain length and can be formed by bulk, solvent, suspension or emulsion polymerization.

Copolymers derived from mixtures of any of the aforedescribed vinyl monomers either with themselves or with other commonly used polymerizable monomers are used herein. Such copolymers possess a wide range of properties and can be formulated to fit individual needs.

5

10

15

20

25

One class of resins especially useful herein are the alkyd resins. Such resins are polyesters of polyhydroxyl alcohols and polycarboxyl acids chemically combined with various drying, semi-drying and non-drying oils in different proportions. Thus, for example, the alkyd resins are made from polycarboxylic acids such as phthalic acid, maleic acid, fumaric acid, isophthalic acid, succinic acid, adipic acid, azelaic acid, sebacic acid as well as from anhydrides of such acids, where they exist. The polyhydric alcohols which are reacted with the polycarboxylic acid include glycerol, trimethylolethane, trimethylolpropane, pentaerythritol, sorbitol, mannitol, ethylene glycol, diethylene glycol and 2,3-butylene glycol.

The alkyd resins are produced by reacting the polycarboxylic acid and the polyhydric alcohol together with a drying, semi-drying or non-drying oil in proportions depending upon the properties desired. The oils are coupled into the resin molecule by esterification during the manufacturing and become an integral part of the polymer. The oil is fully saturated or predominately unsaturated. The fully saturated oils tend to give a plasticizing effect to the alkyd, whereas the predominately unsaturated oils tend to crosslink and dry rapidly with oxidation to give more tough and durable alkyd resins. Suitable oils include coconut oil, fish oil, linseed oil, tung oil, castor oil, cottonseed oil, safflower oil, soybean oil, and tall oil. Various proportions of the polycarboxylic acid, polyhydric alcohol and oil are used to obtain alkyd resins of various properties.

Also useful herein are polyester type resins. As conventionally used and as used herein, the term "polyester" is applied to resins which contain no oil or fatty acid modification. That is, while the above-described alkyd resins are in the broadest sense polyester type resins, they are oil-modified and thus not generally considered a polyester

14

resin. The polyesters are of two kinds. The more common are the unsaturated polyesters derived from unsaturated polyfunctional acids and polyhydric alcohol. These polyesters are essentially linear in structure. Maleic acid and fumaric acid are the usual unsaturated acid components. Commonly used polyhydric alcohols are ethylene glycol, propylene glycol, diethylene glycol, dipropylene glycol, butylene glycol, glycerol, trimethylol propane, pentaerythritol and sorbitol. Oftentimes a saturated acid will be included in the reaction to provide desirable properties. Examples of saturated acids include phthalic acid, isophthalic acid, adipic acid, azelaic acid, sebacic acid and the anhydrides thereof where they exist. The saturated polyesters are derived from saturated or aromatic polyfunctional acids, preferably dicarboxylic acids, and mixtures of polyhydric alcohols having an average hydroxyl functionality greater than 2.

Useful acrylic resins are the polymerized ester derivatives of acrylic acid and methacrylic acid. The resins contain the units

$$CH_3$$
 CH_2
 CH_2
 CH_2
 CH_2
 CH_2
 CH_2
 CH_3
 CH_2
 CH_3
 CH_2
 CH_3
 CH_2
 CH_3
 CH_3

respectively. The esters are formed by the reaction of acrylic or methacrylic acid with suitable alcohols, e.g., methyl alcohol, ethyl alcohol, propyl alcohol, butyl alcohol and 2-ethylhexyl alcohol. Generally speaking, the larger the alcohol portion of the ester, the softer and more flexible the resultant resin. Also, generally speaking, the methacrylate esters form harder films than the corresponding acrylic esters. Monomers such as styrene, vinyl toluene, vinyl chloride and vinylidine chloride may be reacted with the acrylic and methacrylic esters so as to produce resins with excellent properties.

Thermosetting acrylic resins are normally low molecular weight copolymers made from 2 and sometimes 3 monomers. One of the monomers is an acrylic compound

BNSDOGID: <WO

03002674A1 | >

5

10

15

containing pendant reactive groups such as carboxyl, hydroxyl or amide. Another is an acrylic ester. The third monomer is usually a styrene type monomer such as styrene itself, vinyl toluene, methyl styrene or ethyl styrene. The proportions of the three components in the polymerization procedure are varied depending on the products in which the copolymer will be used.

5

10

15

20

25

Another class of film-forming resins useful in the tinting compositions herein is the amino resins commonly referred to as aminoplasts. The amino resins are made by the reaction of an amine with an aldehyde. The more important and preferred amines are urea and melamine. The aldehyde component comprises from 1 to 4 carbon atoms, with formaldehyde being the preferred aldehyde. Films of varying properties can be obtained by changing the proportions of the amine and aldehyde and by adding various chemically-reactive materials during the resin formation. Oftentimes, a lower alcohol, especially butanol, is added during the resin formation to impart desirable properties to the amino resin.

The phenolic resins commonly referred to as phenoplasts are also useful film-formers in the context of the subject invention. The phenolic resins are obtained by the condensation of phenol or substituted phenols with aldehydes. The monohydric phenols such as phenol, cresol and xylanol are the most important since they are readily available and relatively inexpensive. Phenol is the most preferred monohydric phenol. Polyhydric phenols such as resorcinol can also be used herein. Formaldehyde is the preferred aldehyde used in the production of the phenolic resins. Other aldehydes which are also useful include acetaldehyde, butyraldehyde and furfuraldehyde. The preferred phenolic resin is produced by the condensation of phenol and formaldehyde.

Different cellulose derivatives useful herein include nitrocellulose, cellulose acetate, cellulose acetate butyrate and ethyl cellulose. These film-forming materials are well known and are commercially available in varying degrees of substitution and molecular weight. Nitrocellulose is the preferred cellulose derivative.

16

Amide resins found to be useful include those polymers made by condensing adiamine with a dibasic acid. They are characterized by recurring amide groups, -- CONH--, as an integral part of the main polymer chain. Examples of diamines used in producing the polyamide resins include ethylenediamine, diethylenetriamine and hexamethylenediamine. The carboxylic acids are the preferred dibasic acids and include adipic acid, sebacic acid, succinic acid, glutaric acid and azelaic acid.

Another class of film-formers used in embodiments of the invention herein are the urethane resins. These are synthetic polymers that may be either thermoplastic or thermosetting. The basic polymeric unit is —RNHCOOR—. The R groups can be the same or different and can contain other reactive groups, for example, a second -NCO group, a second -OH group, etc. Typically, a polyhydric alcohol is reacted with a polyisocyanate to produce the urethane resin. Useful polyhydric alcohols include ethylene glycol, propylene glycol, butylene glycol, glycerol, trimethylolpropane and hexane triol. Many variations are possible.

15

5

10

The urethane coating can be made from a one-pack or two-pack system. The one-pack urethane contains either an isocyanate prepolymer or a blocked isocyanate. The prepolymers are prepared by the reaction of excess diisocyanate with a polyhydric alcohol. Blocked isocyanates contain no free isocyanate groups and are relatively inactive at room temperature. At elevated temperatures, the compounds dissociate and undergo reactions typical of isocyanates. Typical blocking agents are phenols, thiols, tertiary alcohols and secondary aromatic amines. In a two-pack coating system, typically a polyester polyol and an isocyanate (or isocyanate prepolymer for safety reasons) are mixed at the time of application and applied immediately.

25

20

In certain embodiments, the solvent used in the tinting composition is a mixture of straight and branched chain hydrocarbons and a polydimethylsiloxane. In some embodiments, the solvent used in the tinting composition is a mixture of straight and branched chain hydrocarbons and a mixture of polydimethylsiloxanes. In certain embodiments, the solvent used in the tinting composition is a mixture of straight and branched chain hydrocarbons and a mixture of polydimethylsiloxanes having varying

17

viscosities. In one such embodiment, a mixture of polydimethylsiloxanes having viscosities of 50 cst, 100,000 cst, and 8,000,000-12,000,000 cst is used. In other such embodiments, a mixture of polydimethylsiloxanes having an overall viscosity of 50,000-60,000 cst is used. In other embodiments, the solvent used in the tinting composition is a straight chain hydrocarbon, a branched chain hydrocarbon, or a polydimethylsiloxane.

5

10

15

20

25

Suitable pigments for use in preparing the tinting composition include, but are not limited to, iron oxide, titanium dioxide, mica, calcium sodium borosilicate, organic constituent containing pigments, inorganic constituent containing pigments, fluorescent constituent containing pigments, phosphorescent constituent containing pigments, luminescent constituent containing pigments, ultraviolet absorbing pigments, mineral constituent containing pigments, amorphous glass oxides, polyacrylates, metal flakes, reflective pigments, and mixtures thereof. The selection of the pigment coloration also varies based upon the color choice and intensity of color one wishes to be able to impart onto the surface. In certain embodiments, the pigment is an inorganic constituent containing pigment. A preferred pigment is mica coated with titanium dioxide or iron oxide such as Mearlin® Hi-Lite Sparkle Green, Mearlin® Hi-Lite Sparkle Blue, Mearlin® Hi-Lite Sparkle Gold, Mearlin® Hi-Lite Sparkle Red, Mearlin® Sparkle Russet, Exterior Mearlin® Sparkle, Mearlin® Sparkle Brass, Mearlin® Sparkle Bronze, Mearlin® Sparkle Copper, Mearlin® Hi-Lite Super Green, Mearlin® Hi-Lite Super Blue, Mearlin® Hi-Lite Super Red, Mearlin® Plum, Mearlin® Kiwi Green, Mearlin® Adobe Green, Mearlin® Firemist Red, and Mearlin® Firemist Blue made by Engelhard Corporation, Microlith Yellow made by Ciba Specialty Chemicals, and Helicon® HC made by Wacker. Additional pigments include, but are not limited to, pigments sold by Engelhard Corporation such as the Mearlin® Magnapearls & White Pearlescents, the Mearlin® White Pearlescents, the Mearlin® Hi-Lite Interference Colors, the Mearlin® Dynacolor Luster Pigments, the Mearlin® "Non-Metallic" Metallic Colors, the Mearlin® Santa Fe Colors, the Mearlin® Micro Pearls, the Mearlin® Sparkle Colors, the Mearlite® Industrial Grades, the Mearlin® Firemist pigments, and the Mearlin® Vegetable Colors.

18

Generally, the pigment is present in the tinting composition in an amount from about 0.001 to about 60 wt.% of the tinting composition. In certain embodiments, the pigment is present in an amount from about 0.1 to about 50 wt.% of the tinting composition. In certain other embodiments, the pigment is present in an amount from about 0.1 to about 25 wt.% of the tinting composition.

In addition to a solvent and a pigment, the tinting composition may further include one or more of the following components: thickener flow modifiers, corrosion inhibitors, thickeners, surface tension reducers, silica, emulsifiers, fragrance, release agents, wetting agents, ultraviolet additives, and other additives to add or enhance the desired properties of the tinting composition. The use of these optional components and the desired amounts of each component is based upon the desired properties of the tinting composition such as the degree of coloration and the viscosity of the tinting composition; the shelf life of the tinting composition; the desired scent of the tinting composition; and the means used to apply the tinting composition to the surface. In some embodiments, reflective materials or retroreflective materials are used in place of or in addition to a pigment. Any reflective or retroreflective materials or particles may be used. For example, U.S. Patents No. 4,915,935; 4,983,458; and 5,939,716 disclose various reflective particles which may be used in embodiments of the invention, the disclosures of the patents are incorporated by reference herein in their entirety. U.S. Patents No. 6,066,384 and 5,998,525 disclose various retroreflective particles or fillers which may be used in embodiments of the invention. The disclosures of the patents are incorporated by reference herein in their entirety.

In some embodiments, silica is added to the tinting composition to aid in suspending the pigment within the tinting composition. The silica that is selected for use in the tinting composition can be used either alone or in combination in varying proportions with other silica. Examples of suitable silica which can be used in the present invention include, but are not limited to, fumed silica such as Aerosil® R202 made by Degussa-Hůls and CAB-O-SIL® TS720 made by Cabot Corporation, amide powder, rheology additives, clays, surfactants, cellulose gums, carbomers, precipitated

25

-5

10

15

19

silica such as Zeothix 265 made by J.M. Huber Corporation, and mixtures thereof although other silica may also be used. A preferred silica used in embodiments of the invention is fumed silica. Generally, silica is present in the tinting composition in an amount from about 0.001 to about 30 wt.% of the tinting composition. In certain embodiments, the silica is present in an amount from about 0.01 to about 15 wt.% of the tinting composition. In certain other embodiments, the silica is present in an amount from about 0.1 to about 10 wt.% of the tinting composition.

5

10

15

20

25

In some other embodiments, an emulsifier is added to the tinting composition to aide in holding the components of the tinting composition into solution, to thicken the suspension of the tinting composition, and to serve as a dispersant for the pigment. It is desired to use an emulsifier in the tinting composition where the solvent is water in order to thicken the suspension of the tinting composition. Suitable emulsifiers include, but are not limited to, nonionic, cationic, anionic and amphoteric surfactants, block polymers, block copolymers, acrylics, primary and secondary emulsifiers, and mixtures thereof although other emulsifiers may also be used. The emulsifier that is selected for use in the tinting composition can be used either alone or in combination in varying proportions with other emulsifiers. Generally, an emulsifier is present in the tinting composition in an amount from about 0.001 to about 20 wt.% of the tinting composition. In certain embodiments, the emulsifier is present in an amount from about 0.1 to about 15 wt.% of the tinting composition.

Any emulsifying agent or emulsifier may be used in embodiments of the invention. Suitable emulsifying agents include, but are not limited to the following compounds identified by the INCI names: 7-dehydrocholesterol; Abietic acid; Acetylated glycol stearate; Avocadamide DEA; Bentonite; Beta-sitosterol; C₁₂₋₁₃ alcohols; C₁₂₋₁₅ alcohols; C₁₂₋₁₆ alcohols; C₁₅₋₁₈ glycol; C₂₂₋₂₄ pareth-33; C₉₋₁₁ alcohols; Calcium carrageenan; Carbomer; Carboxymethyl hydroxyethylcellulose; Carboxymethyl hydroxypropyl guar; Carrageenan; Cellulose gum; Cera microcristallina; Ceresin; Cetearyl alcohol; Cocamide; Cocamide DEA; Cocamide MEA; Cocamide MIPA; Cocamidopropyl lauryl ether; Dihydroxyethyl cocamine oxide; Ethylene/ acrylic acid

20

copolymer; Ethylene/VA copolymer; Glycol cetearate; Hydrogenated C₁₂₋₁₈ triglycerides; alcohol; Hydrogenated tallow amide; Hydroxybutyl lanolin Hydrogenated methylcellulose; Hydroxyethyl ethylcellulose; Hydroxyethylcellulose; Hydroxypropyl guar; Hydroxypropyl methylcellulose; Hydroxypropylcellulose; Isopropyl ester of PVM/MA copolymer; Lanolinamide DEA; Lauryl alcohol; Maltodextrin; Methoxy PEG-22/dodecyl glycol copolymer; Methylcellulose; Microcrystalline cellulose; Montmorillonite; Myristyl alcohol; Oxyquinoline sulfate; Ozokerite; Palm kernelamide DEA; Palm kernelamide MEA; Palm kernelamide MIPA; Palmamide DEA; Palmamide MEA; Palmamide MIPA; Peanutamide MEA; Peanutamide MIPA; Pectin; PEG-115m; PEG-14m; PEG-20m; PEG-23m; PEG-2m; PEG-45/dodecyl glycol copolymer; PEG-5m; PEG-7m; PEG-90m; PEG-9m; Pentadecyl alcohol; Polyacrylic acid; Polyvinyl acetate; PVM/MA PVP; copolymer; Potassium carrageenan; alginate; Potassium PVP/dimethylaminoethylmethacrylate copolymer; PVP/VA copolymer; Saccharated lime; Scleroglucan; Sclerotium gum; Sodium acrylate/vinyl alcohol copolymer; Sodium olefin/maleic acid copolymer; Sodium carboxymethyl dextran; Sodium Sodium sulfate; Sodium polymethacrylate; Sodium cellulose carrageenan; polynaphthalenesulfonate; Sodium polystyrene sulfonate; Soyamide DEA; Stearic acid; Stearyl alcohol; Styrene/MA copolymer; Synthetic beeswax; Synthetic candelilla wax; Synthetic carnauba; Synthetic japan wax; Synthetic wax; Tallow amide; Tallowamide DEA; Tallowamide MEA; Tetrasodium etidronate; Tridecyl alcohol; and Xanthan gum.

In some embodiments, a fragrance is added to provide a scent to the tinting composition. Any fragrance known in the art which imparts the desired scent to the tinting composition can be used. The fragrance that is selected for use in the tinting composition can be used either alone or in combination in varying proportions with other fragrances. Generally, a fragrance is present in the tinting composition in an amount from about 0.0001 to about 15 wt.% of the tinting composition. In certain embodiments, the fragrance is present in an amount from about 0.01 to about 5 wt.% of the tinting composition.

25

5

10

15

In some other embodiments, an ultraviolet additive is added to protect the tinting composition from the sunlight. Examples of suitable ultraviolet additives include, but are not limited to, free radical scavengers, antioxidants, hindered amines, antistatic agents, and mixtures thereof although other ultraviolet additives may also be used. The ultraviolet additive that is selected for use in the tinting composition can be used either alone or in combination in varying proportions with other ultraviolet additives. Generally, an ultraviolet additive is present in the tinting composition in an amount from about 0.001 to about 10 wt.% of the tinting composition. In certain embodiments, the ultraviolet additive is present in an amount from about 0.01 to about 5 wt.% of the tinting composition.

As described above, a number of ingredients or components may be present in the tinting composition in accordance with embodiments of the invention. The following table shows the weight percentages for each component. It should be recognized that these ranges are merely preferences and that not all listed components need to be present in the tinting composition. A composition falling outside the preferred ranges is still within the scope of the invention described and claimed herein. The weight percentages are based on the total weight of the tinting composition. It should be recognized that the ingredients or components of the tinting composition can be added in any order to prepare the tinting composition.

20

5

10

15

TABLE 1

Component	Range (wt.%)	Preferred Range (wt.%)
Solvent	0.1-99.9%	50-99.9%
Pigment	0.001-60%	0.1-50%
Silica	0-30%	0.01-15%
Emulsifier	0-20%	0.1-15%
Fragrance	0-15%	0.01-5%
Ultraviolet Additive	0-10%	0.01-5%
Film Former	0.01-25%	0.5-15%

Although various tinting compositions are disclosed, it should be understood that a coating composition comprising the components of a tinting composition except a

pigment may be prepared in accordance with embodiments of the invention. The coating composition can be used for decoration or protection of articles. Any component suitable for formulating a tinting composition can be used in a composition. In some embodiments, a coating composition comprises at least one film-former and a carrier or solvent. Pigments can also be used if opacity or color is desired. Various coating additives can also be used to control the physical properties of the composition. Suitable film-formers can be one of the following polymeric materials: thermoset rubbers (crumb rubbers); thermoplastic rubbers or elastomers such as block copolymers whose hard segments in the polymer molecule consist of styrene and the soft segments isoprene, butadiene, ethylene-propylene, or ethylene-butylene; polyvinyl butyral; or mixtures thereof. Suitable carriers can be one of the following solvents: mineral oils; volatile hydrocarbons; ketones; esters; ethers; chlorohydrocarbons; fluorohydrocarbons; alcohols; glycols; or mixtures thereof. It should be understood that any of the components disclosed herein may be used in a coating composition. A coating composition may become a tinting composition when certain pigments are added.

The following are preferred embodiments for basic coating compositions. Composition A is a basic formulation. Composition B includes pigmentation. Composition C includes water, an emulsifier and pigmentation. The ranges given are for uses with a wide variety of release agents, solvents, emulsifiers, pigments.

20

5

10

15

TABLE 2 BASIC FORMULAS

	A wt %	B wt %	C wt%
Polymer	1-25	1 25	1 - 25
Release Agent	0.1 - 10	0.1 - 10	0.1 - 10
Solvent	5 - 90	5 - 50	50 - 90
Emulsifier			0.1 - 10
Water			5 – 50
Pigment		0.1 - 40	0.1 - 10

The release agent in the above compositions may be a silicone, an organo silicone or a long chain fatty acid ester or polyester. The solvent may be a 1:1 volume of

toluene/mixed heptanes, mineral spirits or blends thereof. The emulsifier may be a polyacrylate or a nonionic surfactant. The polymer may be a styrene/butylene copolymer which may further be functionalized. The above compositions may further include a film former, a wetting agent (surfactant), neutralizers.

5

The following examples are presented to illustrate various embodiments of the invention. All numerical values are approximate numbers. The specific details in each example should not be construed to limit the invention as otherwise described and claimed herein. The following tables show various tinting compositions made in accordance with embodiments of the invention. The amount of each component added to each tinting composition is provided in weight percent of the total composition.

10

15

EXAMPLE 1

INGREDIENT	SUPPLIER	CHEMICAL NAMES	PRINCIPLE FUNCTION	wt.%
Luviskol k-60	BASF	Polyvinylpyrrolidone (MW 350,000)	Film former	5
PhosFlex	Azko Nobel	Tributoxyethyl phosphate	Flow modifier	ì
Zonyl	Dupont	Fluorosurfactant	Surface tension reduction	0.1
Pigment-dye				0.5
IPA		Isopropanol	Solvent	93.4
Total				

Note: the polymer Luviskol k-60 can be replaced by other polymers: PVP-VA I-335 (available from ISP Technologies); PVP/Vinyl Acetate; Gantrez ES-435 (available from ISP Technologies); or Butyl ester of PVM/MA copolymer.

EXAMPLE 2

INGREDIENT	SUPPLIER	wt.%
Penreco Ultima Petrolatum	Penreco	2
AP-200 Silicone	Wacker	2
Igepal DAP-9	Witco	0.2
Dupont R-900 TiO2	Dupont	15
15% Kraton G-1652 in Cypar 9		70.8
Kraton	Shell	10.62
Cypar 9	Shell	60.18
Dye-pigment	Day-Glow	10
Total Ingredients		100

24 EXAMPLE 3

This example is preferably used for inorganic surfaces such as glass or metals.

Ingredient	Wt.%
Crumb rubber	8.5%
Tetrachloroethelene	76%
Silicone oil	0.5%
White pigment	5%
Color pigment	10%

EXAMPLE 4

This example forms a clear and flexible coating and can be used for various surfaces.

Ingredient	Wt.%
Thermoplastic rubber such as Styrene-	5-30%
ethylene/butylenes-styrene block copolymer	
Mineral oil	10-80%
Volatile organic solvent	0-50%

EXAMPLE 5

This example forms a color coating.

Ingredient	Wt.%
Thermoplastic rubber such as Styrene-	20%
ethylene/butylenes-styrene block copolymer	
Mineral oil	55%
di-isobutyl ketone	10%
White pigment	5%
Color pigment	10%

EXAMPLE 6

This example forms a clear and flexible coating and can be used for various surfaces.

Ingredient	Wt.%
polyvinyl butyral	5-30%
Low alcohol solvent	45-95 %
Release agent	0-2%
Antifoaming/defoaming agent	0-2%
White pigment	0-10%
Color pigment	0-10%

BNSDOCID: <WO ____03002674A1_I_>

The following examples form a clear and flexible coating and can be used for various surfaces.

Component	Example 7	Example 8	Example 9
Polymer	50	50*	50
Surfactant	0.6	0.6	0.6
Emulsifier	0.15	0.15	0.3
Water	49.025	49.025	48.65
Neutralizer	0.225	0.225	0.45

^{*} also includes 2% release agent

The polymer in the above examples can be represented by a 10% solution of Kraton G-1652 from Kraton Polymers (Houston, TX) in mineral spirits. The surfactant of the above example can be represented by Igepal CO-530 from Rohm and Haas (Philadelphia, PA). The emulsifier can be represented by Pemulen 1622 from Noveon, Inc. (Cleveland, OH). The neutralizer can be represented by AMP-95 by 2-amino-2-methyl-1-propanol that is 95% active from Angus Chemical (Buffalo Grove, IL). The release agent can be represented by Ketjenlube 165 from Akzo Noble (Chicago, IL).

The following examples show the variety of polymers, surfactants, solvents and release agent components that can be utilized to produce a clear and flexible coating which can be used for various surfaces. Wetting agents were added to the compositions to improve the application of the compositions.

	Polymer (A, B or C)	Wetting Agent (A, B, or C)	Solvent (A, B, or C)	Release Agent
Example 10	5 (A)	0.5 (A)	94.5 (A)	0
Example 11	7.5 (B)	0.5 (A)	92 (A)	0
Example 12	7.5 (B)	0.5 (A)	91.0 (B)	1.0
Example 13	7.5 (B)	0.5 (A)	91.0 (B)	1.0
Example 14	7.5 (B)	0.5 (A)	91.0 (A)	1.0
Example 15	10 (C)	0.5 (A)	89.5 (A)	
Example 16	7.5 (B)	0.5 (A)	91.0 (C)	1.0
Example 17	7.5 (B)	0.5 (B)	91.0 (B)	1.0
Example 18	7.5 (B)	0.5 (B)	91.0 (B)	1.0
Example 19	7.5 (B)	0.5 (B)	91.0 (B)	1.0
Example 20	7.5 (B)	0.5 (C)	91.0 (A)	1.0
Example 21	7.5 (B)	0.5 (C)	91.0 (A)	1.0

15

DEICHMONTH AME

02000074A+ 1 ~

10

The polymers in the above examples can be represented by (A) Kraton G-1651, (B) Kraton G-1652 and (C) Kraton G-1725 all from Kraton Polymers (Houston, TX). The wetting agents in the above examples can be represented by (A) Monazoline O from UNIQEMA (New Castle, DE), (B) Miramine CC from Aventis (Bridgewater, NJ) and (C) Monazoline T from UNIQEMA (New Castle, DE). The release agent in the above examples can be represented by Ketjenlube 165 by Akzo Nobel (Chicago, IL). The solvents in the above examples can be represented by (A) 45:50 chlorobenzene/VM&P naphtha, (B) 45:50 perchloroethylene/VM&P naphtha, and (C) 2-chloro-toluene/VM&P naphtha. Most of the compositions produced a clear formulation. The compositions with wetting agent (A) appeared to have an improved wetting effect on tire samples.

The following examples have a cross-linking agent added to the coating composition. The cross-linking agent promotes adhesion within the formula, not of the formula to the surface.

	Polymer	Wetting	Solvent	Release	Cross-linking
	(A, B, C or D)	Agent	(A, B, C, D)	Agent	Agent (A, B, or C)
Example 22	7.5 (A)	0	90.5 (A)	1.0	1.0 (A)
Example 23	7.5 (A)	0	90.5 (A)	1.0	1.0 (B)
Example 24	7.5 (B)	0	90.5 (A)	1.0	1.0 (A)
Example 25	5.0 (D)	0	93.0 (A)	1.0	1.0 (A)
Example 26	5.0 (A)	0	93.0 (B)	1.0	1.0 (C)
Example 27	7.5 (C)	0	90.0 (C)	1.0	2.0 (C)
•			0.5 (D)		
Example 28	15.0 (C)	0	80.0 (C)	2.0	2.0 (C)
*			1.0 (D)	220000000000000000000000000000000000000	
Example 29	11.25 (C)	0	85.0 (C)	1.5	1.5 (C)
			0.75 (D)		
Example 30	7.5 (C)	0.5	89.5 (C)	1.0	1.0 (C)
-			0.5 (D)		

15

5

10

The polymers in the above examples can be represented by (A) Kraton G-1652, (B) Kraton G-1726, (C) Kraton G-1726M all from Kraton Polymers (Houston, TX) and (D) TUFTEC M1913 from Marubeni America Corporation (NY, NY). The release agent in the above examples can be represented by Ketjenlube 165 by Akzo Nobel (Chicago,

5

10

15

20

IL). The solvents in the above examples can be represented by (A) 30:70 perchloroethylene/heptanes, (B) perchloroethylene, (C) 1:1 toluene/heptanes, and (D) Wacker 1000 cs Silcone from Wacker Chemical (Adrian, MI) and mixtures thereof. The wetting agents in the above examples can be represented by Surfadone LP-300 from International Specialty Products (Wayne, NJ). The cross-linking agents in the above examples can be represented by (A) octadecenyl succinic acid (ODSA), (B) Dimer acid diglycidyl ester, (C) and EDENOL 9232 (ESBO) from Henkel KgaA (Hayward, CA). The above compositions were tested for opacity of the solution and adhesion to tires.

The following example is for a tire coating composition that may be packaged in an aerosol can.

EXAMPLE 31

Ingredient	Wt.%
Polymer A	4.0
Polymer B	3.5
Release Agent	1.0
Solvent A	0.5
Solvent B	90.0
Cross-linking Agent	1.0

The polymers in the above examples can be represented by (A) Kraton G-1726M, and (B) Kraton G-1652 both from Kraton Polymers (Houston, TX). The release agent in the above example can be represented by Ketjenlube 165 by Akzo Nobel (Chicago, IL). The solvents in the above examples can be represented by (A) 1:1 toluene/heptanes and (B) Wacker 1000 cs Silcone from Wacker Chemical (Adrian, MI) and mixtures thereof. The cross-linking agents in the above examples can be represented by EDENOL 9232 (ESBO) by Henkel KGaA (Hayward, CA).

The following examples have a pigment added to the coating composition. The pigments impart a color to the surface it is applied to. The surfaces it may be applied to include tires, glass, and paint.

28

The state of the s	Polymer	Surfactant	Solvent	Release	Pigment
	and Solvent	(A,B,C,D,E,F,		Agent	
		G,H, or I)		(A or B)	
Example 32	94.0	2.0 (A)	2.0 (A)	, , , , , , , , , , , , , , , , , , ,	2.0
Example 33	94.0	2.0 (B)	2.0 (A)		2.0
Example 34	94.0	2.0 (C)	2.0 (A)		2.0
Example 35	94.0	2.0 (D)	2.0 (A)		2.0
Example 36	94.0	2.0 (E)	2.0 (A)		2.0
Example 37	92.0	2.0 (F)	2.0 (A)	2.0 (A)	2.0
Example 38	94.0	2.0 (G)	2.0 (A)		2.0
Example 39	94.0	2.0 (H)	2.0 (A)		2.0
Example 40	94.0	2.0 (I)	2.0 (A)		2.0
Example 41	94.0	2.0 (I)		2.0 (A)	2.0
Example 42	90.0	2.0 (A)	2.0 (A)	2.0 (A)	2.0

The polymer and solvent component in the above examples can be represented by 10% Kraton G-1652 in a 1:1 volume of toluene and heptanes. The release agents in the above examples can be represented by (A) Super Petrolatum from Penreco (Houston, TX) or (B) OZOKERITE 77Y Wax from Frank B. Ross Co. (Jersey City, NJ). The solvent (A) in the above examples can be represented by Wacker 1000 cs Silcone from Wacker Chemical (Adrian, MI). The surfactants in the above examples can be represented by (A) Witco Varonic Q202 from Crompton Corporation (Greenwich, CT), (B) Igepal CO-530 from Rohm and Haas (Philadelphia, PA), (C) Ethomeen 0/15 from Akzo Nobel (Chicago, IL), (D) Chemeen 18-2 from Chemtex (East Providence, RI), (E) Chemeen 18-5 from Chemtex (East Providence, RI), (F) Monazoline O from UNIQEMA (New Castle, DE), (G) Chemal OA-5 from Chemax, Inc. (Greenville, SC), (H) Surfadone LP-300 from International Specialty Products (Wayne, NJ), and (I) Witconate R-1059 from Thornley Co. (Wilmington, DE). The pigment may be represented by SBT-353GN from Silverline Manufacturing Co. (Tamaqua, PA). The pigments may also be any color.

Examples which have pigments, may further have titanium dioxide (< 5 microns) added to the composition. The titanium oxide provides opacity to the composition and enhances the pigmentation.

10

5

29

While the invention has been described with respect to a number of limited embodiments, variations and modifications exist. For example, although the invention is described with reference to vehicles, the application of the invention is not limited only to use on vehicles. U.S. Patents No. 4,339,365; 5,340,390; 5,707,697; and 6,187,233B1 disclose various coating compositions and method of using same. The various components and method of use may be used in embodiments of the invention with or without modifications. All of the preceding patents are incorporated by reference herein in their entirety. Additional objects on which the tinting composition may be used include, but are not limited to, bicycles, motorcycles, campers, motor homes, recreational vehicles, trucks, roller blades, boats, and canoes. Also, other surfaces described herein can be tinted regardless of whether the surface is in or on a vehicle or other objects specifically referenced. The appended claims intend to cover all such variations and modifications as falling within the scope of the invention as described herein.

What is claimed is:

ANADOMOTIAL For

5

30

- 1. A tinting composition for tinting a surface comprising:
 - a solvent in the amount from about 50 to about 99 wt%;
 - a film former in the amount from about 0.01 to about 25 wt%; and

5

- a pigment in the amount from about 0.001 to about 60 wt% selected from the group consisting of iron oxide, titanium dioxide, mica, calcium sodium borosilicate, organic constituent containing pigments, inorganic constituent containing pigments, fluorescent constituent containing pigments, phosphorescent constituent containing pigments, luminescent constituent containing pigments, ultraviolet absorbing pigments, mineral constituent containing pigments, amorphous glass oxides, polyacrylates, metal flakes, reflective pigments, and mixtures thereof.
- 2. The tinting composition of claim 1, wherein the surface is a polymeric substrate.
 - 3. The tinting composition of claim 2, wherein the polymeric substrate is selected from the group consisting of rubber, engineered plastic, and vinyl.
- 4. The tinting composition of claim 2, wherein the polymeric substrate is selected from the group consisting of the surface of a tire, dashboard, bumper, steering wheel, mudflap, and seat.
 - 5. The tinting composition of claim 1, wherein the surface is selected from the group consisting of metal, glass, fiberglass, plastic, and paint.
- 6. The tinting composition of claim 1, wherein the solvent is selected from the group consisting of branched or straight chained aliphatic hydrocarbons, synthetic hydrocarbons, alcohols, water, organic solvents, silicone fluids, paraffinic solvents, volatile solvents, and mixtures thereof.
- 7. The tinting composition of claim 1, wherein the solvent is a silicone fluid comprising polydimethylsiloxane.

- 8. The tinting composition of claim 7, wherein the polydimethylsiloxane has a viscosity ranging from about 0.65 cst to about 12,000,000 cst.
- 9. The tinting composition of claim 1, wherein the solvent is a mixture of straight and branched chain hydrocarbons.
- 5 10. The tinting composition of claim 1, wherein the solvent is a mixture of straight and branched chain hydrocarbons and a silicone fluid comprising polydimethylsiloxane.
 - 11. The tinting composition of claim 1, wherein the pigment is an inorganic constituent containing pigment.
 - 12. The tinting composition of claim 1, wherein the pigment is selected from the group consisting of mica coated with titanium dioxide and mica coated with iron oxide.
 - 13. The tinting composition of claim 1, further comprising silica.

- 14. The tinting composition of claim 13, wherein the silica comprises from about 0.001 to about 60 wt.% of the tinting composition.
- 15. The tinting composition of claim 13, wherein the silica is selected from the group consisting of fumed silica, amide powder, rheology additives, clays, surfactants, cellulose gums, carbomers, precipitated silica, and mixtures thereof.
 - 16. The tinting composition of claim 1, further comprising an emulsifier.
 - 17. The tinting composition of claim 16, wherein the emulsifier comprises from about 0.001 to about 20 wt.% of the tinting composition.
- 18. The tinting composition of claim 17, wherein the emulsifier is selected from the group consisting of nonionic, cationic, anionic and amphoteric surfactants, block polymers, block copolymers, acrylics, primary and secondary emulsifiers, and mixtures thereof.

- 19. The tinting composition of claim 1, further comprising fragrance.
- 20. The tinting composition of claim 19, wherein the fragrance comprises from about 0.0001 to about 20 wt.% of the tinting composition.
- 21. The tinting composition of claim 1, further comprising an ultraviolet additive.
- 5 22. The tinting composition of claim 21, wherein the ultraviolet additive comprises from about 0.001 to about 10 wt.% of the tinting composition.
 - 23. The tinting composition of claim 1, wherein the pigment comprises from about 0.1 to about 50 wt.% of the tinting composition.
- 24. The tinting composition of claim 1, wherein the solvent is a mixture of straight and branched chain hydrocarbons and a silicone fluid comprising polydimethylsiloxane and the pigment is an inorganic constituent containing pigment.
 - 25. The tinting composition of claim 24, wherein the pigment is selected from the group consisting of mica coated with titanium dioxide and mica coated with iron oxide.
 - 26. The tinting composition of claim 1, further comprising silica, emulsifier, fragrance and an ultraviolet additive.
 - 27. The tinting composition of claim 26, wherein the silica comprises from about 0.001 to about 30 wt.%, the emulsifier comprises from about 0.001 to about 20 wt.%, the fragrance comprises from about 0.0001 to about 15 wt.% of the tinting composition, and the ultraviolet additive comprises from about 0.001 to about 10 wt.% of the tinting composition.
 - 28. A tinting composition for tinting a surface comprising:
 - a solvent selected from the group consisting of branched or straight chained aliphatic hydrocarbons, synthetic hydrocarbons, alcohols, water, organic solvents, silicone fluids, paraffinic solvents, volatile solvents, and mixtures thereof;

15

5

10

15

20

a polyvinylpyrrolidone vinyl acetate copolymer;

a pigment selected from the group consisting of iron oxide, titanium dioxide, mica, calcium sodium borosilicate, organic constituent containing pigments, inorganic constituent containing pigments, fluorescent constituent containing pigments, phosphorescent constituent containing pigments, luminescent constituent containing pigments, ultraviolet absorbing pigments, mineral constituent containing pigments, amorphous glass oxides, polyacrylates, metal flakes, reflective pigments, and mixtures thereof; and

a silica selected from the group consisting of fumed silica, amide powder, rheology additives, clays, surfactants, cellulose gums, carbomers, precipitated silica, and mixtures thereof.

- 29. The tinting composition of claim 28, wherein the surface is a polymeric substrate.
- 30. The tinting composition of claim 29, wherein the polymeric substrate is selected from the group consisting of rubber, engineered plastic, and vinyl.
- 31. The tinting composition of claim 29, wherein the surface is selected from the group consisting of the surface of a tire, dashboard, bumper, steering wheel, mudflap, and seat.
- 32. The tinting composition of claim 28, wherein the surface is selected from the group consisting of metal, glass, fiberglass, paint, and plastic.
- 33. The tinting composition of claim 28, wherein the solvent is a mixture of straight and branched chain hydrocarbons and a silicone fluid comprising polydimethylsiloxane, the pigment is selected from the group consisting of mica coated with titanium dioxide and mica coated with iron oxide, and the silica is fumed silica.
- 34. The tinting composition of claim 28, wherein the tinting composition further comprises a fragrance.
- 25 35. A method of tinting a surface comprising:

34

selecting a surface for tinting;

mixing a solvent in the amount from about 50 to about 99.9 wt %, a film former in the amount from about 0.01 to about 25 wt%, and a pigment in the amount from about 0.001 to about 60 wt% selected from the group consisting of iron oxide, titanium dioxide, mica, calcium sodium borosilicate, organic constituent containing pigments, inorganic constituent containing pigments, fluorescent constituent containing pigments, phosphorescent constituent containing pigments, luminescent constituent containing pigments, ultraviolet absorbing pigments, mineral constituent containing pigments, amorphous glass oxides, polyacrylates, metal flakes, reflective pigments, and mixtures thereof to form a tinting composition; and

applying the tinting composition to the surface.

- 36. The method of claim 35, further including the step of drying the tinting composition.
- 37. The method of claim 35, wherein the step of mixing further includes adding a silica.
- 38. The method of claim 35, wherein the step of mixing further includes adding an emulsifier.
- 39. The method of claim 35, wherein the step of mixing further includes adding a fragrance.
- 40. The method of claim 35, wherein the step of mixing further includes adding an ultraviolet additive.
- 41. The method of claim 35, wherein the step of applying is accomplished by spraying the tinting composition onto the surface.
 - 42. The method of claim 35, wherein the step of applying is accomplished by one of the steps selected the group consisting of sponging, dabbing, painting and wiping the tinting composition onto the surface.

5

10

35

- 43. The method of claim 35, wherein the step of applying is accomplished by applying an adhesive which contains the tinting composition onto the surface and removing the adhesive.
- 44. The method of claim 35, wherein the surface is a polymeric substrate.

10

0200002444 | -

DESCRIPTION AND

- 5 45. The method of claim 44, wherein the polymeric substrate is selected from the group consisting of rubber, engineered plastic, and vinyl.
 - 46. The method of claim 44, wherein the polymeric substrate is selected from the group consisting of the surface of a tire, dashboard, bumper, steering wheel, mudflap, and seat.
 - 47. The method of claim 35, wherein the surface is selected from the group consisting of metal, glass, fiberglass, paint, and plastic.
 - 48. The method of claim 35, wherein the solvent comprises from about 50 to about 99.9 wt.% and the pigment comprises from about 0.1 to about 50 wt.% of the tinting composition.

Interconal Application No PCT/US 02/20870

A. CLASSIFICATION OF SUBJECT MATTER IPC 7 C09D5/36 C09D11/00 C08J7/04 A61K7/42 C03C17/32 C09D131/0 B05D7/00 B44C1/17 C09D131/04 According to International Patent Classification (IPC) or to both national classification and IPC **B. FIELDS SEARCHED** Minimum documentation searched (classification system followed by classification symbols) C09D C08J B05D B44C A61K C03C IPC 7 Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched Electronic data base consulted during the international search (name of data base and, where practical, search terms used) EPO-Internal, PAJ C. DOCUMENTS CONSIDERED TO BE RELEVANT Citation of document, with indication, where appropriate, of the relevant passages Relevant to claim No. Category ° 1,6,11, EP 0 931 538 A (L'OREAL) 23,35,36 28 July 1999 (1999-07-28) example 1 EP 0 691 387 A (SAKURA COLOR PROD CORP) 1,6,11, X 23,35, 10 January 1996 (1996-01-10) 36,42 claims 1-4; examples 7,8 page 3, line 53 X EP 0 435 951 A (SNELL & WILCOX LTD) 1,6,11, X 13-15, 10 July 1991 (1991-07-10) 23, 35-37, examples 1,2 Patent family members are listed in annex. Further documents are listed in the continuation of box C. Special categories of cited documents: *T* later document published after the international filing date or priority date and not in conflict with the application but *A* document defining the general state of the art which is not cited to understand the principle or theory underlying the considered to be of particular relevance invention earlier document but published on or after the International *X* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to filing date Involve an inventive step when the document is taken alone *L.* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the citation or other special reason (as specified) document is combined with one or more other such docu-*O* document referring to an oral disclosure, use, exhibition or ments, such combination being obvious to a person skilled other means in the art. *P* document published prior to the international filing date but "&" document member of the same patent family later than the priority date claimed Date of mailing of the international search report Date of the actual completion of the international search 11/11/2002 1 November 2002 Authorized officer Name and mailing address of the ISA European Patent Office, P.B. 5818 Patentlaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nl. Schueler, D Fax: (+31-70) 340-3016

Form PCT/ISA/210 (second sheet) (July 1992)

C.(Continual	ion) DOCUMENTS CONSIDERED TO BE RELEVANT	
Category °	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Х	EP 0 176 141 A (HOLMQVIST OLLE) 2 April 1986 (1986-04-02) table 1	1,6,11, 23,35,36
X	DE 43 19 669 A (BASF AG) 13 January 1994 (1994-01-13)	1,6,11, 12,35, 36,41, 45,47,48
	examples	
X	DE 30 20 381 A (STAEDTLER FA J S) 3 December 1981 (1981-12-03)	1,6,11, 16-18, 23,35, 36,42, 47,48
	page 6	
χ	GB 405 986 A (FREDERICK WILLIAM SKIRROW; SAMUEL WHYTE) 12 February 1934 (1934-02-12)	1,6,9, 11,23, 35,36, 41,47
	claims; examples 8,10-12	
Х	GB 2 104 908 A (PELIKAN AG) 16 March 1983 (1983-03-16) examples	1,6,11, 23,35
X	FR 2 782 918 A (OREAL) 10 March 2000 (2000-03-10) claims 1,14,15,27,28; examples	1,6,11, 23,35,36
χ	FR 2 659 011 A (OREAL) 6 September 1991 (1991-09-06)	1,6,11, 15-18, 28,35, 42,48
	example 6	
X	JP 2000 204307 A (FUJI SHIKISO KK) 25 July 2000 (2000-07-25)	1,6,11, 16-18, 21,23, 35,36, 38,41, 42,48
	paragraph '0022! - paragraph '0028!; claims	
X	US 2 346 624 A (STRAUS EDWIN G) 11 April 1944 (1944-04-11) page 2, right-hand column, line 1 - line 25	1,6,11, 23,35,36
X	US 2 368 161 A (GUSTAVE RUBNER) 30 January 1945 (1945-01-30) page 3, left-hand column, line 45 -page 4, left-hand column, line 23	1,6,11, 23,35,36
	-/	
1		

Internal Application No PCT/US 02/20870

C.(Continua	ation) DOCUMENTS CONSIDERED TO BE RELEVANT	
Category °	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 3 283 036 A (LARSON CRAIG W) 1 November 1966 (1966-11-01)	1,6-8, 11,23, 35,36, 41,47
	example 7	
χ	US 3 466 181 A (ALLISON KENNETH C ET AL) 9 September 1969 (1969-09-09)	1,6,11, 12,23, 35,36, 42,47
Α .	column 1, line 58 -column 2, line 29 column 3, line 34 - line 37; claim 1	28-34
Χ	US 3 823 205 A (ZIMMT W) 9 July 1974 (1974-07-09)	1,6-8, 11,23, 35,36
X	example 1 US 4 244 862 A (HANDA MINORU ET AL) 13 January 1981 (1981-01-13)	1,6,9, 11, 16-18, 23,35, 36,38, 41,47,48
	examples 4,7	
X	US 4 544 578 A (DUANE JR GEORGE P) 1 October 1985 (1985-10-01)	1,6,11, 13-20, 23,35, 36,39, 42,44-48
	example 4	
X	US 4 569 887 A (COWLES RICHARD A) 11 February 1986 (1986-02-11)	1,6,11, 13-18, 23,35, 36,38,41
	examples 1,2	
X	US 4 962 139 A (LO SALVADOR R) 9 October 1990 (1990-10-09) example 8	1,6,11, 23,35,36
X X	column 2, line 60 column 5, line 65 -column 6, line 20	7,8 16-18, 38,41, 44-48
X	WO 95 29286 A (GEN ELECTRIC CO PLC) 2 November 1995 (1995-11-02) examples	1,6,11, 23,35,36

Interion No PCT/US 02/20870

C.(Continu	ation) DOCUMENTS CONSIDERED TO BE RELEVANT	
Category °	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	WO 99 22588 A (RAADGEVEND CHEMIEBUREAU RSB V ; BERTELS ANTONIUS FRANCISCUS MA (NL)) 14 May 1999 (1999-05-14) claim 1; example 1 page 4, line 8 - line 15	1,6,11, 23,35,36 44,45,47
X X X X	page 4, line 34 -page 9, line 14 claim 9 page 12, line 14	41,42 11 16-18
P,X	US 2002/033117 A1 (INOUE HIROSHI ET AL) 21 March 2002 (2002-03-21)	1,6, 11-19, 23,35,36
	tables 1,2	
Р,Х	US 2002/013398 A1 (ARAI KENJIRO ET AL) 31 January 2002 (2002-01-31) table 3	1,6,11, 23,35,36
P,X	US 2002/006481 A1 (MORRIS BARRINGTON A) 17 January 2002 (2002-01-17) paragraph '0026! - paragraph '0039!; claims 12-18	1,6,11, 21-23,35
P,X	EP 1 153 594 A (CODEN CO LTD) 14 November 2001 (2001-11-14)	1,6,11, 28,35, 42,48

Form PCT/ISA/210 (continuation of second sheet) (July 1992)

International application No. PCT/US 02/20870

INTERNATIONAL SEARCH REPORT

Box I Observations where certain claims were found unsearchable (Continuation of item 1 of first sheet)
This International Search Report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:
1. Claims Nos.: because they relate to subject matter not required to be searched by this Authority, namely:
2. X Claims Nos.: 2-5 because they relate to parts of the International Application that do not comply with the prescribed requirements to such an extent that no meaningful International Search can be carried out, specifically: see FURTHER INFORMATION sheet PCT/ISA/210
3. Claims Nos.: because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).
Box II Observations where unity of invention is lacking (Continuation of item 2 of first sheet)
This International Searching Authority found multiple inventions in this International application, as follows:
1. As all required additional search fees were timely paid by the applicant, this International Search Report covers all searchable claims.
2. As all searchable claims could be searched without effort justifying an additional fee, this Authority did not invite payment of any additional fee.
3. As only some of the required additional search fees were timely paid by the applicant, this International Search Report covers only those claims for which fees were paid, specifically claims Nos.:
4. No required additional search fees were timely paid by the applicant. Consequently, this International Search Report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:
Remark on Protest The additional search fees were accompanied by the applicant's protest. No protest accompanied the payment of additional search fees.

Form PCT/ISA/210 (continuation of first sheet (1)) (July 1998)

FURTHER INFORMATION CONTINUED FROM PCT/ISA/ 210

Continuation of Box I.2

Claims Nos.: 2-5

Present claims 1,6-48 relate to an extremely large number of possible products and methods. In fact, the claims contain so many options, variables, possible permutations and provisos that a lack of clarity (and/or conciseness) within the meaning of Article 6 PCT arises to such an extent as to render a meaningful search of the claims impossible. Consequently, the search has been carried out for those parts of the application which do appear to be clear (and/or concise), namely those mentioned in the description at pages 23-28, especially the concentrations given in examples 1-3, 5, 32-42:

a composition characterised by

a) nonaqueous solvent in an amount of >50 wt%,

b) film former in an amount of from 5 to 25 wt%,

c) pigment in an amount of from 2 to 15 wt%

characterized that the %sum a+b+c is not greater than 100.

The claims 2-5, formulated in the form of dependent claims linked to the composition claim 1, describe substrate conditions whereby the text of the claims lack any information regarding the step of converting the fluid composition into a substrate covered/treated therewith. In fact, these claims lack clarity within the meaning of Article 6 PCT to such an extent as to render a meaningful search of the claims impossible

The applicant's attention is drawn to the fact that claims, or parts of claims, relating to inventions in respect of which no international search report has been established need not be the subject of an international preliminary examination (Rule 66.1(e) PCT). The applicant is advised that the EPO policy when acting as an International Preliminary Examining Authority is normally not to carry out a preliminary examination on matter which has not been searched. This is the case irrespective of whether or not the claims are amended following receipt of the search report or during any Chapter II procedure.

Information on patent family members

Internal Application No
PCT/US 02/20870

				PCT/US	02/208/0
Patent document cited in search repor	-t	Publication date		Patent family member(s)	Publication date
EP 0931538	A	28-07-1999	FR BR DE	2773067 A1 9805784 A 69805950 D1	02-07-1999 11-04-2000 18-07-2002
			DE	69805950 T2	26-09-2002
			EP	0931538 A1	28-07-1999
			JP	2983537 B2	29-11-1999
			JP	11246358 A 6203806 B1	14-09-1999 20-03-2001
parent private opinio dilato biggo quint ant is some 21 new brings quinti anti-	Mai sphilips Shiphage Salam _{a Manager} wasters States Matter States	B ليسبب - فيدوان جاهدات هندسام ولموسط «الأمراد واسبول ويتموه كباراته فياوان والأماثة والماثية والمساور الموس	US 	شناست فيست والمدارة والسابة والمدارة والمراث والمراث والمراث والمراث والمراث والمراث والمدارة	province having tall, and analysis grappine Market Arrand systems channel thomas Milled D ^{esc} ions through grapper
EP 0691387	A	10-01-1996	JP	3304612 B2	22-07-2002
			JP DE	7331155 A 69506954 D1	19-12-1995 11-02-1999
			DE	69506954 T2	29-07-1999
			EP	0691387 A2	10-01-1996
			US	5698614 A	16-12-1997
EP 0435951	A	10-07-1991	DE	68923289 D1	03-08-1995
			DE	68923289 T2	07-03-1996
			EP	0435951 A1 1002246 A1	10-07-1991 07-08-1998
			HK AT	124589 T	15-07-1995
			AU	4348289 A	18-04-1990
			WO	9003705 A1	05-04-1990
			JP	2913507 B2	28-06-1999
			JP US	4500892 T 5166792 A	13-02-1992 2411-1992
designs appears can see Fishing Private brights biddle appeals frames delicate de	_{report} animal jar sep galgal, _{Miles V} entere anima marro m	ديسون حيومية والمثلثة فالمراج عديدة والمدار والمدارة والمدارة والمدارة والمدارة والمدارة والمدارة والمدارة والمدارة	وهوامه وبالمالية والمالية والمالية والمالية والمالية	MANUAL PROPERTY PROPERTY AND ADMINISTRATION OF PERSONS AND ADMINIS	T dimmire himmire diagrap distribit himite deligas philific himite white smilet himite deligas deligas deligas
EP 0176141	A	02-04-1986	AT	45109 T	15-08-1989 14-09-1989
			AU AU	588325 B2 5830086 A	10-12-1987
			CA	1234941 A1	05-04-1988
			DE	3571943 D1	07-09-1989
			DK	390785 A ,B,	25-03-1986
			EP	0176141 A2	02-04-1986
			ES	547225 DO	01-05-1988
			ES	8802289 A1	16-07-1988
			FI NO	853290 A ,B, 853603 A ,B,	25-03-1986 25-03-1986
			SE	8404754 A	25-03-1986
			US	4670321 A	02-06-1987
DE 4319669	Α	13-01-1994	DE	4319669 A1	13-01-1994
DE 3020381	A	03-12-1981	DE	3020381 A1	03-12-1981
GB 405986	A	12-02-1934	FR	790736 A	26-11-1935
GB 2104908	A	16-03-1983	DE	3214305 A1	13-01-1983
			AT	384585 B	10-12-1987
			AT	151782 A 1170511 Al	15-05-1987 10-07-1984
			CA CH	656351 A5	30-06-1986
			FR	2508383 A1	31-12-1982
	•		IT	1195793 B	27-10-1988
			JP	1013438 B	06-03-1989
			JP	1526684 C	30-10-1989
			JP	58029694 A	21-02-1983
			ÜS	4515489 A	07-05-1985

Form PCT/ISA/210 (patent family annex) (July 1992)

Information on patent family members

Interconal Application No PCT/US 02/20870

					101/03	02/208/0
	it document search report		Publication date		Patent family member(s)	Publication date
FR 27	782918	A	10-03-2000	FR AT BR CN DE DE EP ES JP	2782918 A1 205703 T 9904272 A 1248432 A 69900299 D1 69900299 T2 0997139 A1 2165226 T3 2000086487 A	10-03-2000 15-10-2001 26-09-2000 29-03-2000 25-10-2001 18-04-2002 03-05-2000 01-03-2002 28-03-2000
FR 26	659011		06-09-1991	FR AU AU CA DE DE EP EO JP US	2659011 A1 105171 T 635630 B2 7452491 A 2046600 A1 69101889 D1 69101889 T2 0471054 A1 2052376 T3 9112793 A1 2623168 B2 4505469 T 5925337 A	06-09-1991 15-05-1994 25-03-1993 18-09-1991 02-09-1991 09-06-1994 22-12-1994 19-02-1992 01-07-1994 05-09-1991 25-06-1997 24-09-1992 20-07-1999
JP 2	000204307	A	25-07-2000	NONE		
US 2	346624	Α	11-04-1944	NONE		
US 2	368161	A	30-01-1945	NONE	Annual States Service Service Service Columbia Santas Service	
US 3	283036	A	01-11-1966	BE DE FR GB NL SE	631378 A 1246916 B 1363438 A 993376 A 292048 A 319256 B	10-08-1967 12-06-1964 26-05-1965 12-01-1970
US 3	466181	Α	09-09-1969	NONE		
US 3	8823205	A	09-07-1974	BE CH DE FR GB MY NL SE	689966 A 475328 A 1669017 A1 1500730 A 1124690 A 19469 A 6616448 A 325354 B	22-05-1967 15-07-1969 25-03-1971 03-11-1967 21-08-1968 31-12-1969 23-05-1967 29-06-1970
US 4	1244862	A	13-01-1981	JP JP JP JP AU AU DE FR	1257142 C 52094329 A 59032501 B 1196927 C 53121031 A 58025354 B 509653 B2 2979577 A 2749238 A1 2385777 A1	29-03-1985 08-08-1977 09-08-1984 21-03-1984 23-10-1978 26-05-1983 22-05-1980 26-04-1979 05-10-1978 27-10-1978

Form PCT/ISA/210 (patent family annex) (July 1992)

Information on patent family members

Internal Application No PCT/US 02/20870

Det	ant donument		Publication	y allufunin finin 1111. Phase deposits age to the term	Patent family	-	Publication
	ent document in search report		date		member(s)		date
US	4244862	Α		GB	1564380 /		10-04-1980
				NL	7712116 /	4 ,B,	03-10-1978
US	4544578	A	01-10-1985	NONE			
US	4569887	Α	11-02-1986	US	4539360	4	0309-1985
 บร	4962139	A	09-10-1990	US	4745012	Α	17-05-1988
				CA	1307367	-	08-09-1992
				CA	1324339	A2 	16-11-1993
WO	9529286	Α	02-11-1995	US	5510415	A	23-04-1996
				AT	171489		15-10-1998
				AU	2144995		16-11-1995
				CA	2187914		02-11-1995
				DE	69505002	-	29-10-1998
				DE	69505002		18-02-1999
				EP	0778907		18-06-1997
				WO	9529286		02-11-1995
	nan yanan dagan dan ya maya Jamah dalah katan dan dalah d	***************************************	فيهيه وويدو ويسهم ومشاه فمشاه فسيدة فميهن ويميين يانتيني وماندي ومانتين وعايسر يوواتك أد	JP	10501006		27-01-1998
WO	9922588	A	14-05-1999	NL	1007433		04-05-1999
				AT	199205		15-03-2001
				BR	9814107		03-10-2000
				CA	2302738		14-05-1999
				DE	69800536		29-03-2001
				DE	69800536		27-09-2001
				DK	999736	_	11-06-2001
				EP	0999736		17-05-2000
				ES	2156447 2001521944		16-06-2001 13-11-2001
				JP WO	9922588		14-05-1999
	darife listenti eligeny divindo kanar norza menen septen porto darda darife (ileak)	يزدة شاقلك و'نامو هسته باللبي	opp styres t-midd dat was littled statute differe fidence amon writer tourned littles between "and	هاهان مادم الأدارية ويبيد الاسام الاسام الاسام	فالمرا المامية	ad annum mengaan kumbuk kebulah diseban kebuntu bilan	THE BUILDING STREET, ST
US	2002033117	A1	21-03-2002	JP	2002249694	A	06-09-2002
US	2002013398	A1	31-01-2002	JP	2001271033	A	02-10-200
US	2002006481	A1	17-01-2002	NONE		may differe chique bules over over some some	nyar 'ya yak jamu'y kuthish Milishiji aktauy assar k assas assas assas assas kumu katalik Kutala k
EP	1153594	A	14-11-2001	JP	2002003335	Α	09-01-200
•	· · · · · · · · · · · · · · · · · · ·			CN	1328812	A	02-01-200
				EP	1153594		14-11-200
				US	2002006422	Mar	17-01-200